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INDUSTRIAL WASTE WATER RECYCLING USING NANO GRAPHENE OXIDE FILTERS

by Yuvarani P

Submission date: 26-May-2021 01:45PM (UTC+0530)

Submission ID: 1594452831

File name: Yuvarani_Paper.docx (1.93M)

Word count: 2090

Character count: 12157

INDUSTRIAL WASTE WATER RECYCLING USING NANO GRAPHENE OXIDE FILTERS

ABSTRACT

Nanomaterials play a vital role in healthcare, electronics, manufacturing industries, biotechnology and security systems. One such material was Graphene and its oxides specifically used for recycling industrial waste water. Graphene, a single layer in honeycomb cross section, providing excellent attentions because of its significant optical, mechanical, physical and optical properties. GO was utilized to decrease the acidic or essential centralization of the mechanical wastewater into reusable water for the modern reason utilizing graphene channels. Graphene channels can be created from the pencil graphite. Graphene have the high goals of separating capacity and graphene was considered as “A definitive RO film” in light of its stronger, thinner and more chemically safe than the polymer layers. Graphene oxide layers were likewise to be used in the desalination plant in replace with the RO membrane.

Index Terms - Graphene Oxide (GO), Hummers Method, FTIR, pH Analysis

1. INTRODUCTION

Nowadays waste water coming from industries will produces major problem to surroundings as well as creates pollution to the environment. Public and government forced the industries to recycle or reduce the waste coming out to stringent standards. So many industries looking forward to implement better and portable treatment technologies. Graphene was obtained from pencils and then oxidation process was carried out in order to scatter the carbon layers with oxygen molecules, then carbon layers are completely separated into single or multilayer by means of reduction [1] , Graphene has possess extremely strong bond not only mechanically but also have high electro migration [2-4,8,13,18,26]. As well as it cannot dispersed in water or any other organic solvent [5], other admirable properties with the improvement of an reasonable, biodegradable and mass GO production, which would be highly significant [6,15] . GO-silicon bi-layer structure experimented for humidity sensing detection which exhibited outstanding humidity sensitivity [7]. Graphene has promised has extended potential research applications in solar cells, corrosion prevention, fuel cells display panels, circuit boards, detection of diseases [9-10,17] and flexible films [25] . The mechanical properties of Polyurethane / Graphene Oxide/Multilayer-Graphene (PU/GO-MG) composites exhibits excellent chemical bonding [11]. Chemically reduced graphene oxide (RGO) revealed highly sensitivity to NO₂ at the room temperature of 200°C [12]. Polypyrrole/Graphene Oxide (PPy/GO) indicate enhanced electrochemical performances because of its flexibility and electro chemical activity [14]. Liquid crystallinity nature of the graphene oxide had well demonstrated [15]. Graphene Oxide (GO) membranes clearly exhibited as excellent membrane for gas separation process, it blocks everything except water vapour [16] Graphene oxide (GO) derived via chemically has used as humidity sensitive coating deposited over quartz crystal microbalances (QCMs) to detect various Relative Humidity at room temperature [22-23]. The capacity for graphene oxide to direct electrons relies upon the measure of oxidization in the compound, just as the strategy for amalgamation. It's the oxidization in the arrangement that aggravates electrical conductivity [20], so while profoundly oxidized graphene oxide would be an extremely poor channel of power, regardless of whether graphene oxide were to be vigorously

decreased, however it would potentially have the capacity to lead power, it would in any case not execute just as high quality graphene monolayers as far as electron versatility. In any case, there are techniques that can be used to check this circumstance, and those are by a procedure frequently alluded to as functionalization, which intends to artificially change a substance in request to build up its properties to suit a particular application. The properties includes dispersibility, Toxicity, Hygroscopicity and filtering ability. The subsequent artificially adjusted graphene could then conceivably turn out to be much increasingly versatile for practically boundless applications. Functionalization of graphene oxide can be done by numerous techniques.

2. MATERIALS AND METHODS: Several methods were used for synthesis of GO, such as Hummer's method, Modified Hummers method, Graphite Oxide reduction method, Graphite intercalation method, electrochemical method, organic synthesis method, carbon nano tubes conversion method, liquid stripping method, solvent thermal method, and arc-discharge method. One of the most commonly used method was Hummer's Method [24]. This method was highly recommended because, the GO yield was high and cost also low compared with other methods.

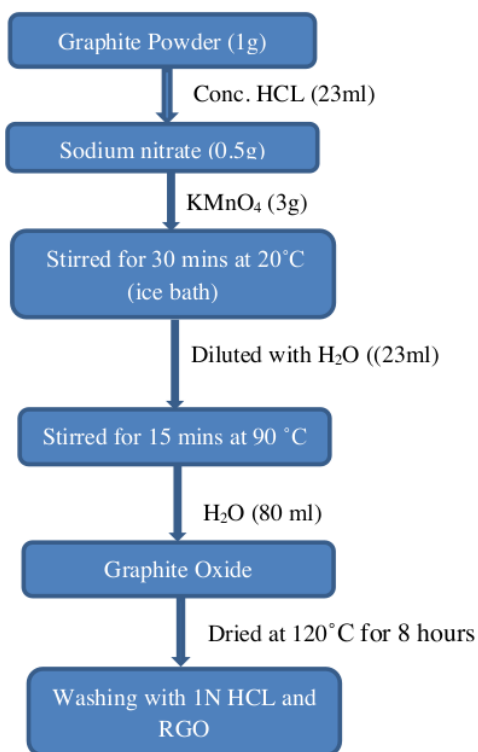


Figure 1. Flow chart of Hummer's method

2.1 Chemicals required: Graphite powder, Concentrated HCL (Hydrochloric acid), Sodium Nitrate, Potassium permanganate, Distilled water, Sulphuric acid, Hydrogen peroxide.

2.2 Hummer's Method of synthesis: Graphene oxide was synthesized by Hummer's Method through oxidation of graphite [24, 27]. The stepwise procedure was mentioned as follows:

1. Graphite powder (1g) and Conc. HCL(23ml) is taken in a beaker and kept in an ice bath at a temperature (0-5°C)
2. Sodium Nitrate (0.5g) is added in a beaker and stirred for 30 mins at 20 °C and KMnO_4 (3g) is added into the beaker
3. After 30 mins the beaker is diluted with H_2O (23ml) and stirred for 15 mins at 90 °C after some time H_2O (80ml) is added to the solution
4. After stirring some time Graphite Oxide is synthesized and then is washed with the 1N HCL (Hydrochloric acid) several times
5. The washed Graphite oxide is dried at 120°C for 8 hours and finally Reduced Graphene Oxide
6. The obtained Graphene Oxide may have some other chemical impurities, after centrifugation and further drying process the final Graphene Oxide Nano filters

GO was analysed using FT-IR analysis for atomic characteristics. The FTIR analysis proved that the sample was converted into a Graphene Oxide compound. The water samples from the industries and other impure water samples were taken and it was analysed and filtered using Graphene Oxide Nano particles. Graphene Oxide allows only the water molecules and filter out various others chemical particles and impurities in the water sample. Graphene possessed the ability to filter out the various micro-organisms, dissolved salts, and various types of impurities in the water or industrial waste water.

3. RESULTS AND DISCUSSION:

The incorporated Graphene oxide (GO) was prepared by Hummer's method and result was revealed by FTIR.



Fig. 2. Mixture at Ice-bath at below 2°C for about 1 hour



Fig. 3 Nano particles settled down after 8 hours



Figure 4. centrifugation at 1500 rpm for 20 minutes

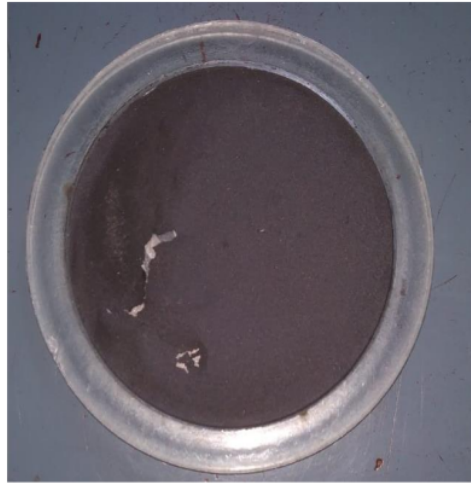
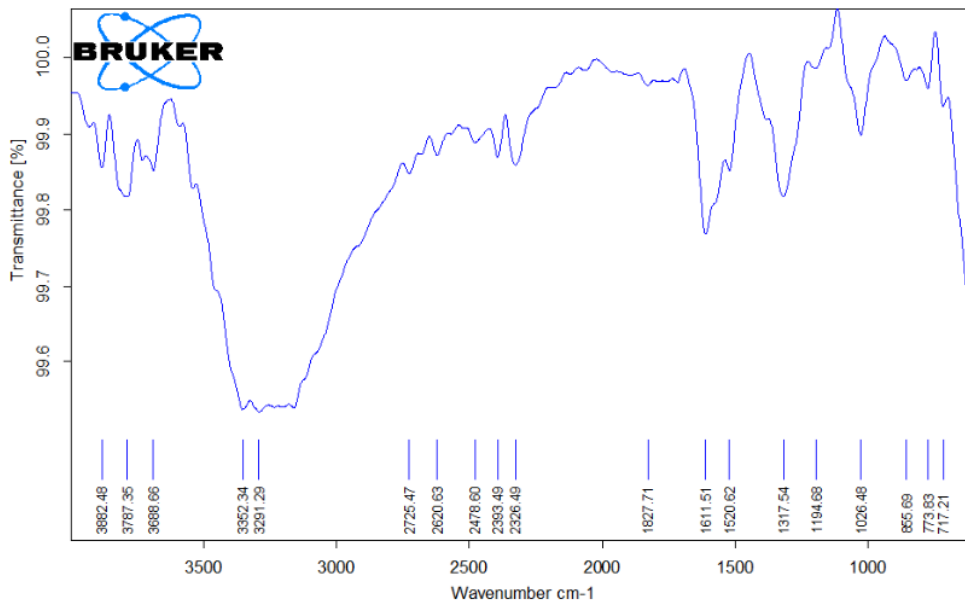


Figure 5. Dried at 120°C for 8 hours

3.1 Fourier Transform Infrared Spectrum Analysis



E:\Consultancy\Prakash\02042019\Sample description.1	Sample description	Instrument type and / or accessory	02-04-2019
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Figure 6. Fourier transform infrared spectra of GO

Fourier transform infrared spectroscopy used to examine the bonding structure. It is a framework grasped to get an infrared scope of maintenance, transmission, and photoconductivity of a solid, gas or liquid. It expects the powers of the pinnacles are specifically identified with the sum of test present [19]. The incorporated GO has a crest at 1081 cm^{-1} which is ascribed to the C-O bond, affirming the nearness of oxide useful gatherings after the oxidation procedure. The crests in the scope of 1630 cm^{-1} to 1650 cm^{-1} demonstrate that the C=C security still stayed previously what's more, after the oxidation procedure. The ingested water in GO is appeared a wide top at 2885 cm^{-1} to 3715 cm^{-1} , contributed by the O-H extend of H_2O particles.

3.2 pH Analysis



Figure 7. Sample Solution pH value



Figure 8. Filter using GO Nano Particles

The sample solution containing some impurities and acidic content was first analysed and pH of that sample solution is measured. Then synthesized GO nano particles was placed over the filter paper and the sample solution to be tested poured over the filter paper.

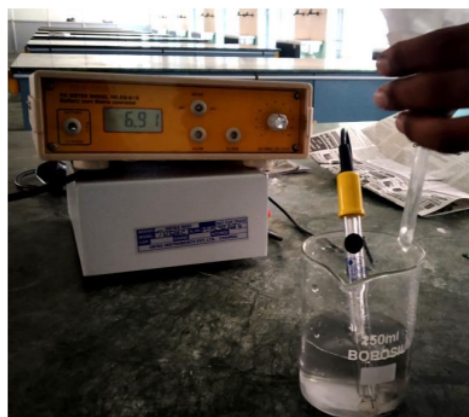


Figure 9. pH value after filtration

GO was produced from pencil graphite so it may have some impurities and it reduces the pH level. On further filtration with the purified GO nano particles layer or paper it will give better reduction in the pH value to around neutral level.

4. CONCLUSION

Nano Graphene Oxide filter paper was obtained by Hummers method. Fourier Transform Infrared Spectrum revealed the presence of C-O and C=C bonds. The graphene Oxide Nano filters is capable of filtering out the impurities and also ability to reduce the pH level of the industrial waste water into a reusable form. Apart from this filtering ability, Graphene Oxide also capable of gas sensing, energy storing ability and current and voltage characteristics. Thus, Graphene was one of the important and future of electronics, electrical and various other important technologies around the world.

5. REFERENCES

1. Paulchamy B, Arthi G and Linges GD. 2015. A Simple approach to Stepwise Synthesis of Graphene Oxide Nanomaterial, *Journal of Nano medicine and Nanotechnology*, 6(253).
2. Leila Shahriary, Anjali A.Athwale. 2014. Graphene Oxide Synthesized by using Modified Hummers Approach, *International Journal of Renewable Energy and Environmental Engineering*,2(1) .
3. Chao Zhou, Sihao Chen, Jianzhong Lou, Zhinchang Chen. 2015. Recent Advances in Graphene Preparation Methods, *Materials Science Forum*, 814: 3-12.
4. Weinchiang R.Chung, Yunfan Zhao, Michael Oye and Cattein Nguyen. 2011. Graphene Synthesis by Thermal –CVD Method, 2011 11th IEEE International Conference on Nanotechnology.
5. Huitao Yu, Bangwen Zhang, Chaoke Bulin and RuinhongLi. 2016. High Efficient Synthesis of Graphene Oxide Based on Improved Hummers’s Method, *Scientific Reports*, 6(1).
6. Xioshan Zhang, Haun Wang, Tianjiao Haung, Lingling Wen, Liya Zhou. 2018. Synthesis of Graphene Oxide through different Oxidation degrees for solar cells, *Materials Research Express*, doi.org/10.1088/2053-1591/aab580
7. Yao Yao, Xiangdong Chen, Huihui Guo, Zuquan Wu and Xiaoyu Li. (2012), Humidity Sensing behaviours of Graphene Oxide – Silicon bi-layer flexible structure, *Sensors and Actuators B*, 161: 1053 – 1058.
8. V.K.Singh, M.K.Petra, M.Manoth, G.S.Gowd, S.R.Vadera and N.Kumar, (2009), In situ Synthesis of Graphane oxide and its composites with iron oxide, *New Carbon Materials*, 24(2):147-152.
9. Edward P.Randviir, Dale A.C.Brownson, and Craig E,Banks. 2014. A decade of graphene research, production, applications and outlook, *Materials Today*, 17(9): 426-432.
10. Hawraa H. Radey, Hadi Z. Al-Sawaad and Moayed N.Khalaf,(2018), Synthesis and Characterization of Novel Nano Derivatives of Graphene Oxide, *Graphene*, 7: 17-29.
11. L.M.Flores, Velez, O.Dominguez, (2018), Graphene-Oxide / Multilayer- Graphene Synthesized from Electrochemically Exfoliated Graphite and its influence on Mechanical Behaviour of Polyurethane Composites, *Material Science Applications*, 9: 565 – 575.
12. Tran Van Khai, Tran Dai Lam, Le Van Thu and Hyoun Woo Kim,(2015), A Two-step method for the preparation of Highly conductive Graphene Film and its Gas Sensing Property, *Materials Sciences and Applications*, 6:963-977.

13. L.M.Manocha, H.L. areja, S.Manocha, (2013), Synthesis of free standing conducting graphene paper by thermal reduction of graphane oxide paper, AIP Advances – Conference Proceedings, 1538(244).
14. Lanyan Li, Keqiang Xia, Liang Li, Songmin Shang, Qingzhong Guo and Guoping Yan, (2012), Fabrication and characterization of free standing polypyrrole and grapheneoxide Nanocomposite paper, Journal of Nanoparticle Research, 14:908.
15. Ji Eun Kim, Tae Hee Han, Sun Hwa Lee, Ju Young Kim, Chi Won Ahm, Je Moon Yun, and Sang Ouk Kim, (2011), Graphene oxide Liquid crystals, Angewandte Chemie, 123: 3099-3103.
16. R.K.Joshi, S.Alwarappan, M.Yoshimura, V. Sahaywalla and Y. Y.Nishina. 2015. Graphene oxide the new membrane material, Applied Materials Today, 1-12.
17. Hai-Sheing Wang, Hong-Xia Chen, Yue-Hong Ma. 2016. The preparation of reduced graphene oxide and its photothermal therapy of gilmas in vivo and in vitro, International Journal of Clinical and Experimental Medicine, 9(6): 10265 – 10272.
18. Adrian Hunt, Dmitriy A, Dikin, Ernst Z. Kurmaev, Teak D. Boyko, Paul Bazylewski, Gap Soo Chang, and Alexander Moewes. 2012. Epoxoide Speciation and Functional Group Distribution in Graphene Oxide paper like materials, Advanced Functional Materials, 22(18):3950 - 3957.
19. L. Guruprasad, V. Krishnakumar and R. Nagalakshmi, 2013. Growth and characterization of semi – organic nonlinear optical crystal: Sodium 2,4-dinitrophenolate monohydrate, Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy 110: 377–382.
20. Sridevi. A, Siva. C, Maheswaran. R and Prasanna Venkatesan G. K. D. 2019. Ga doping improved electrical properties in p-Si /n-ZnO heterojunction diodes, doi.org/10.1007/s10854-019-00891-3 .
21. Md. Sajibul Alam Bhuyan, Md. Nizam Uddin, Md. Maksudul Islam, Ferdaushi Alam Bipasha and Sayed Shafayat Hossain. 2016. Synthesis of graphene, International Journal of Nano Letters, 6 : 65–83.
22. Yao Yao, Xiangdong Chen, Huihui Guo and Zuquan Wu. 2011. Graphene oxide thin film coated quartz crystal microbalance for humidity detection, Applied Surface Science, 257: 7778–7782.
23. F. Perrozzi, S. Prezioso and L. Ottaviano. 2015. Graphene oxide: from fundamentals to applications, Journal of Physics : Condensed Matter, doi:10.1088/0953-8984/27/1/013002.
24. Dmitriy A. Dikin, Sasha Stankovich, Eric J. Zimney, Richard D. Piner, Geoffrey H. B. Dommett, Guennadi Evmenenko, SonBinh T. Nguyen and Rodney S. Ruoff. 2007. Preparation and Characterization of graphane oxide paper, Nature, 448.
25. Gautam Naik and Sridhar Krishnaswamy. 2017 Photoreduction and thermal properties of graphene based flexible films, Graphene, 6: 27-40.
26. Haiqun Chen, Marc B. Müller, Kerry J. Gilmore, Gordon G. Wallace and Dan Li. 2008. Mechanically strong, Electrically Conductive, and Bio-compatible Graphene Paper, Advanced Materials, 20(18): 3557-3561.
27. Ning Cao and Yuan Zhang. 2015. Study of Reduced Graphene Oxide Preparation by Hummers' Method and Related Characterization, Journal of Nanomaterials, 1-5.

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